

PHASE TRANSITION OF GALLIUM ARSENIDE WURTZITE MOLECULES AND NANOCRYSTALS: A DFT STUDY

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This paper presents a detailed analysis of electronic structure and spectroscopic characteristics of GaAs molecules and Wurtzoid nanostructure using density functional theory methods. The results show that the energy gap of GaAs is fluctuated about to the experimental bulk energy gap (1.42 eV). Results of spectroscopic properties including IR and Raman spectrum, reduced mass and force constant are in good agreement with available experimental results. Study of the Gibbs free energy proved the stability phase of GaAs wurtzoids against transition to GaAs diamondoids structure.

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1. Introduction

III–V semiconducting materials like Gallium arsenide (GaAs) in each of nano and bulk forms were always taken under consideration as valuable materials from fundamental and technological [1, 2] viewpoints as a result of their massive potentials for various optoelectronic implementation and due to their possible ability for replacing silicon in micro-electronic components as well, requiring high carrier mobility [3]. Those applications have been based on many preferable characteristics like the vibrational, thermal conductivity, direct band gap, and high electron mobility. A wide range of parameters for band structure of GaAs are famous for a greater accuracy in comparison to any of the other compound semiconductors. Which is particularly accurate of the basic gap of energy with a value of 1.42 eV [4, 5]. In general, those materials have a usual cubic zinc blende (ZB) crystal structure at the ambient conditions in the bulk form [6]. However, recently it has been shown that these compounds can also be grown in other crystal phases which are not possible in the bulk form like the wurtzite (WZ). The wurtzite phase discovery besides the cubic zinc blende phase presents the new way to innovative physics and nano-structure application ideas because of the electronic structure variation with crystal structure [7, 8].

The purpose of this work investigates the electronic characteristics of GaAs molecules, wurtzoids nano-crystals and spectroscopic properties as well as the stability phase transition using density functional theory method.

2. Theory

Binary compounds in periodic table are typically existent in several common phases, including ZB, rock-salt, and WZ structures. Those phases are ones which are projected to undergo growth for the GaAs nano-crystals. Typically, GaAs are available in zinc-blende phase at usual conditions. None-the-less, the compounds can be found in the WZ phase in the case where they are grown on a number of substrate kinds [9, 10]. Wurtzoids can be defined as molecules carrying the characteristics of bulk WZ phase in molecular-nanoscale size. Those molecules are of orientations of preferred axis which are alike (c axis in the WZ phase) and other 2 similar axes (a and b). Those

wurtzoids are created from capped and bundled (3, 0) nano-tubes [11]. The simplest atoms can be represented as in Fig1(c). by elongating this molecule in the c axis, it is possible to obtain the 2nd WZ molecule referred to as the bare wurtzoid-2c as in Fig1 (d) for GaAs wurtzoid-2c. The GaAs wurtzoid-2c stoichiometry is (Ga₁₃As₁₃). Bundling 3 GaAs wurtzoids parallel in the c axis creates GaAs tri-wurtzoid (Ga₂₁As₂₁) as in Fig1 (e) for bare case [11]. Density functional theory (DFT) at B3LYP level (“Becke, threeparameter, Lee-Yang-Parr”) has been utilized, combined with 6-311G** basis sets [12, 14] in this research. All computation are carried out with the use of Gauss09 software [13, 15, 16].

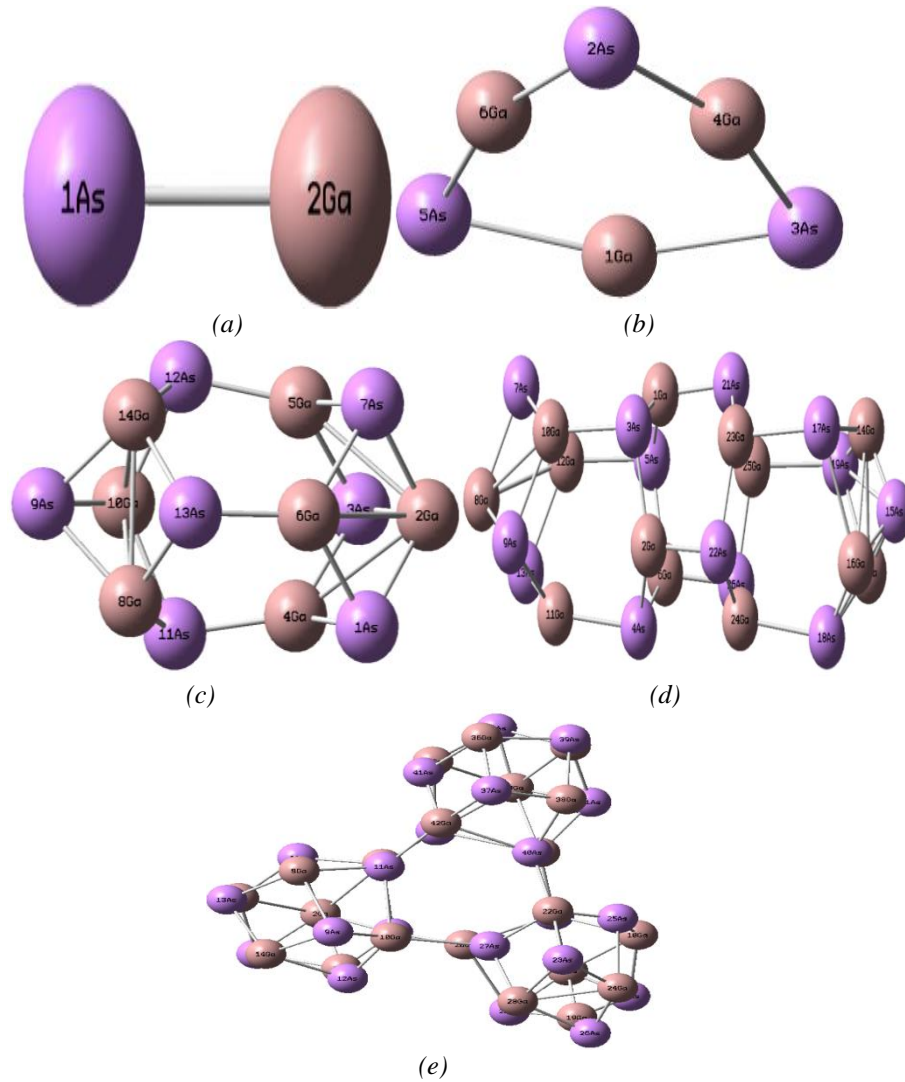


Fig. 1. GaAs Wurtzoid Nanotube Geometrical Optimization Structure: (a) GaAs molecule; (b) Ga₃As₃ Cyclohexane; (c) Ga₇As₇ wurtzoid; (d) Ga₁₃As₁₃ wurtzoid2c and (e) Ga₂₁As₂₁ triwurtzoid.

3. Results and discussions

The energy gap of cases under consideration, which include wurtzoids are shown in Figure (2) also compared to bulk experimental value (1.42 eV) of GaAs wurtzite phase [4,5]. The molecule energy gaps are less compared to the energy gaps of the bulk. The maximum energy gap of the wurtzoids is for the tri-wurtzoids of GaAs, which are equal to 1.71 eV (the entire number of Ga and As atoms = 42), this molecule is almost of a spherical shape with minimal number of the surface dangling bonds explaining its high gap of energy. The elongation of this molecule in c orientation for producing wurtzoid-2c decreases the gap value due to the increased number of the

surface dangling bonds. Continuously elongating those structures in their nano-tubes direction (i.e. the direction of the c parameter) creates GaAs nanowires.

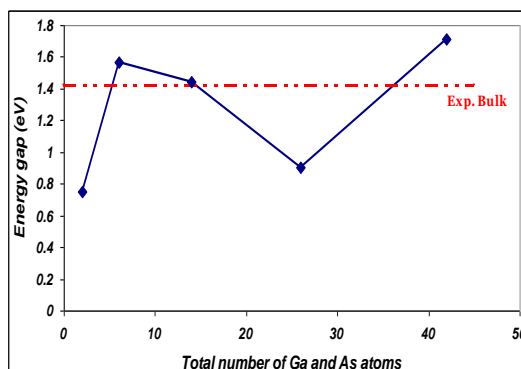


Fig. 2. Energy gap of GaAs wurtzoid in compare with Bulk Experimental Energy Gap = 1.42 eV [4, 5].

Fig. 3 illustrates the bond lengths distribution of the GaAs wurtzoid. There are 3 bond types in those species, which are Ga-Ga, As-As, and Ga-As. The minimal bonds are Ga-As followed by the As-As. The widest bond length distribution is in the (Ga - Ga) bonds.

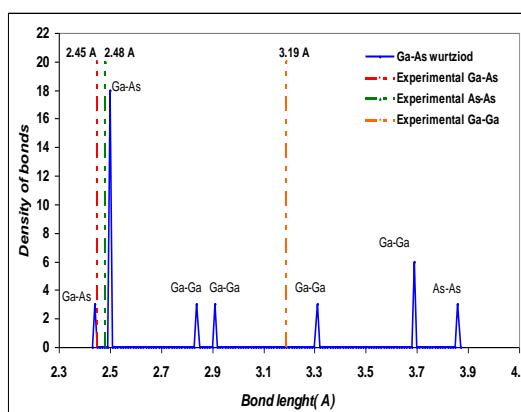


Fig. 3. Density of bond length of Ga₇As₇ (wurtzoid) compare with experimental Ga-As, Ga - Ga, As - As bond lengths are illustrated [14, 15, 17, 18].

Fig. 4 indicates that density of states of wurtzoid (Ga₇As₇) as functions of energy levels. It is found that the band gap (between HOMO (V.B) and LUMO (C.B)) of wurtzoid with exist H-atoms passivated increasing about the experimental value of molecule GaAs of (4.81 eV), while the bare of this structure is 1.44 eV closed to experimental value of GaAs wurtzite (1.42 eV) [4], that means the wurtzoid structure with H-passivated (HP) near to nano-scale lead to increases value of energy gap. Also, the density of states of wurtzoid with exist H-atoms passivated have degenerated of states are lower than of the bare of this structure.

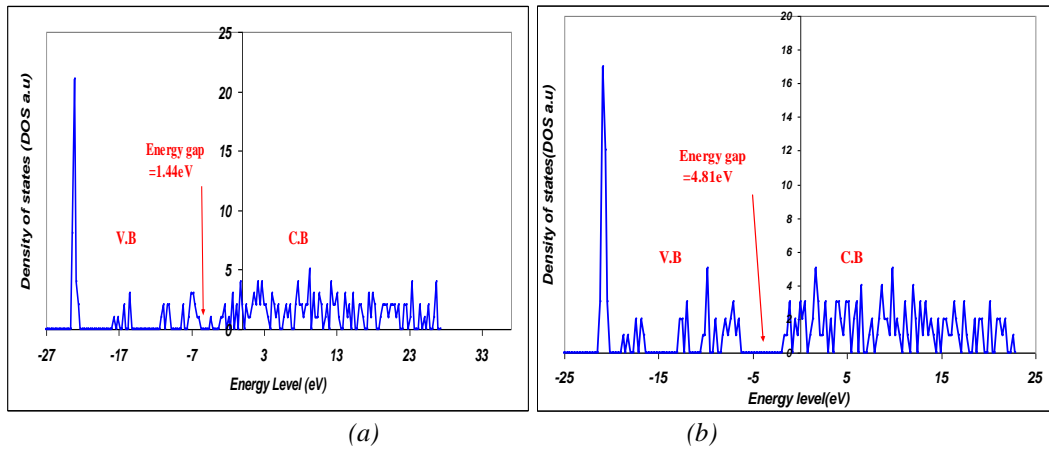


Fig. 4. (a) Density of state (DOS) of Ga₇As₇ bare wurtzoid as a function of energy levels; (b) Density of state of Ga₇As₇H₁₄ HP wurtzoid as a function of energy levels.

Fig. 5 shows the maximal reduced mass mode which is utilized for nanocrystals and molecules for the determination of longitudinal optical (LO) mode at bulk structure for Ga₇As₇ wurtzoid. In this figure appear highest reduced mass mode (HRMM) is equal 282 cm⁻¹ for Ga₇As₇ (wurtzoid) compare with bulk experimental LO mode (285.2 cm⁻¹) [20,21].

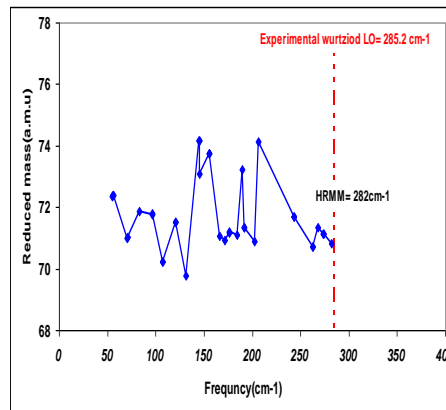


Fig. 5. Reduced mass as a frequency function of Ga₇As₇ wurtzoid.

Fig. 6. shows the vibration of Ga₇As₇ wurtzoid vibrational force constant as a frequency function which appear of parable shape according to relation:

$$v = \frac{1}{2\pi} \sqrt{\frac{k}{\mu'}}$$

where k stands for the force constant, and μ' represents the reduced mass.

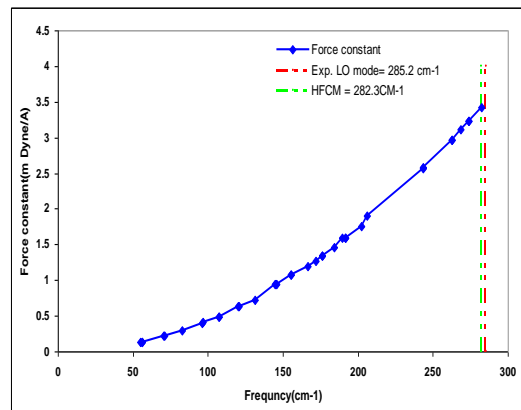


Fig. 6. Force constant as a function of frequency of Ga_7As_7 wurtziod.

Fig. 7 (a) shows IR intensity spectrum, which is observed two regions for Ga_7As_7 wurtziod the first (155-185) cm^{-1} where the center peak at $175.9cm^{-1}$ and the second region center at $341.5 cm^{-1}$. While fig. (7-b) is reveal Raman spectrum which are very low intensity in these two bands. Whereas Bulk GaAs is identified by its infra-red absorption at about 0-300 cm^{-1} vibrational mode peak[22,23].

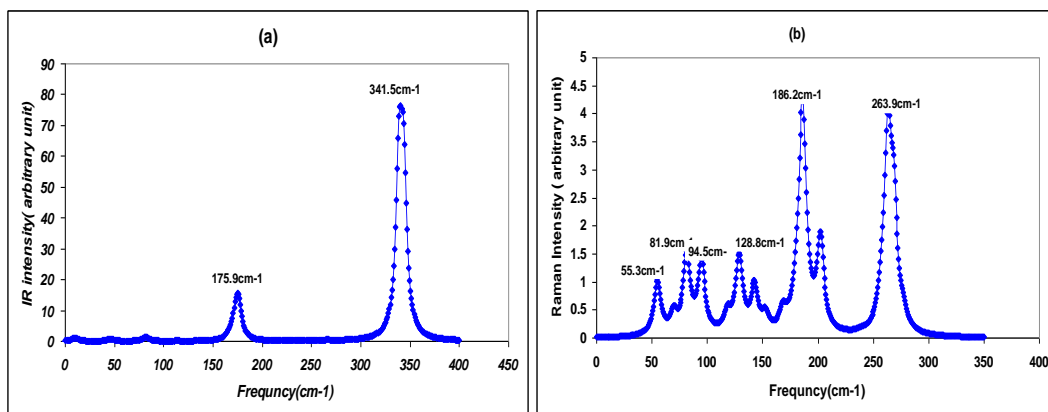


Fig. 7. (a)IR and (b) Raman intensity spectra as a function of frequency of Ga_7As_7 wurtziod.

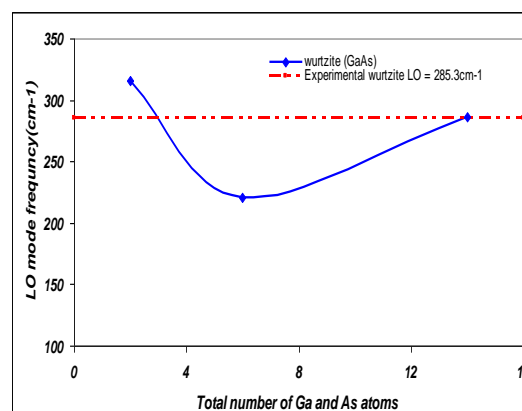


Fig. 8. LO vibrational mode of $GaAs$ wurtziod as a Ga and As atoms' function.

Fig. 8 depicts the LO vibrational mode of GaAs wurtziod as a Ga and As atoms function which decrease the value of LO mode frequency with increasing the number of atoms of GaAs wurtziod. In contrast to the case of the energy gap, the impact of the nano-scale is reducing LO mode frequency value [20,24].

It can be observed that the GaAs wurtziod with (HP) more stable than the GaAs diamantane (HP) due to the Gibbs free energy (G_f) has high (negative) for as mention in Table (1) after take absolute value of them. The diamantane has number of bonds than wurtziod therefore the dangling bonds be more on their surface, than the passivation of H -atom be large, this effect on values of the energy gaps.

Table 1. The relation among several parameters associated GaAs wurtziod (bare and H-passivated) with GaAs diamantane.

Structure	Formula	Energy Gap (E_g)eV	Gibbs Free Energy G_f (eV)	Dangling bonds
Wurtziod GaAs(Bare)	Ga_7As_7	1.44	-11.64	14
Diamantane GaAs(Bare)	Ga_7As_7	1.98	-11.68	20
Wurtziod GaAs (Hp)	$Ga_7As_7H_{14}$	4.81	-12.26	0
Partial Diamantane(HP)	$Ga_7As_7H_{14}$	1.42	-11.97	6

Gibbs Free Energy (G_f) computes as results of subtracting the energy number of clusters from the bulk material. The negative value of (G_f) is spontaneous processes can represent a maximum of work possible [25,26] The absolute values of negative (G_f) give high values of wurtziod and low of diamantane. This belongs to dangling bonds at the fifth column as in Table (1). As a result of these reasons will transform to wurtziod from diamantane at a constant of the pressure and temperature, in all the cases the wurtziod be will more stable with Hp.

The value of energy gap for Diamantane ($Ga_7As_7H_{14}$) is lower than that of wurtziod ($Ga_7As_7H_{14}$) for (H-passivated) case as shown in the table 1, because of the dangling bonds of them.

4. Conclusions

In this work the electronic characteristics of the GaAs wurtziods approach to bulk limits and nano-crystals. Structural parameters like the lengths of the bondshad a valuedistribution where the highest peaks are closer to the optimal values of bulk of those parameters. Results show the energy gap was fluctuated around the experimental bulk gap which generally confined upper wurtziods. Results of vibrational frequencies such as reduced mass and force constant are very converge to their ideal bulk value. In addition to this we have found that the Longitudinal optical (LO) vibrational mode decrease with the increase in the number of GaAs wurtziod atoms and near to ideal bulk value, as well as IR, Raman spectrum intensity are agreement with respect to experimental bulk. The calculations of the Gibbs free energy found that the transformation to the wurtziod phase (HP) were most stable than diamantane structure .

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