

TEMPLATE-FREE SYNTHESIS OF MoSe₂ HOLLOW NANOSPHERES WITH EXCELLENT TRIBOLOGICAL PROPERTIES

M.Q. XUE^{a*}, Z.P. WANG^a, F. YUAN^a, G.S. LUO^a, X.H. ZHANG^b, W. WEI^{c*},
H. TANG^d, C.S. LI^d

^aChangzhou Vocational Institute of Light Industry, Changzhou, Jiangsu Province, 213164, P. R. China.

^bSchool of Mechanical Engineering, Jiangsu University of Technology, Changzhou, Jiangsu Province, 213001, P. R. China

^cSchool of Materials Science and Engineering, Changzhou University, Changzhou, Jiangsu Province, 213164, P. R. China.

^dSchool of Material Science and Engineering, Jiangsu University, Zhenjiang, Jiangsu 212013, P. R. China.

Novel MoSe₂ hollow nanospheres with diameter ~50 nm, consisted of 3~ layered structure MoSe₂ nanosheets, were successfully synthesized by a facile one-step hydrothermal method using selenium powders and ammonium molybdate as raw materials. The as-prepared products were characterized using X-ray diffraction (XRD), energy dispersive spectroscopy (EDS), scanning electron microscopy (SEM) and transmission electron microscopy (TEM); UMT-2 tribotester was used to assess their lubricating effect when used as additives in base oil dispersions. And the topography of worn scars was measured using a BRUKER Contour GT non-contact 3D optical profile testing instrument. Tribological performance showed that the obtained MoSe₂ hollow nanospheres possessed anti-wear and friction-reducing properties as a lubrication additive compared with pure base oil, which will deform and exfoliate and enter into the interface with base oil, then form tribofilm on rubbing surface, enhancing the tribological properties.

(Received October 22, 2016; Accepted February 1, 2017)

Keywords: MoSe₂; Hollow nanospheres; Tribological; Tribo film

1. Introduction

Molybdenum selenide (MoSe₂), as a typical layered transition metal selenide [1–4], in recent years, has garnered a respectable interest for its unusual combination properties and extensive applications in catalysis[5-6], photoluminescence devices [7-8], lithium ion batteries [4], lubricants [9–11], field-effect transistors[12] and so forth. Therefore, it is extraordinarily important to synthesise MoSe₂ with well-controlled dimensionality, size, morphology, and crystal structure for industrial and high technology applications [2,13–16].

To date, several methods, such as hydrothermal and solvothermal method [17, 18], chemical vapor deposition (CVD) [19-21], solid-state reactions [22,23], mechanical exfoliation [24,25], liquid exfoliation [26], and soon have been employed to synthesise MoSe₂nanomaterials. Compared with the other methods, hydrothermal or solvothermal methods can not only induce the formation of well crystallized products with mild synthetic conditions and simple manipulation, but can also control the phase, shape and size of the resultant products simply through adjusting the synthesis conditions such as composition of the solution, pH, temperature, duration, etc.

Hollow nanospheres materials have been regarded as fascinating nanomaterials in the field of batteries [27,28], photocatalysis [29,30], supercapacitor electrodes[31], biosensing [32], gas-sensing[33],and so on. However, MoSe₂ hollow nanospheres have rarely been reported,

*Corresponding author: xuemaq@163.com

therefore, it is still a great challenge to develop a facile and effective process to fabricate MoSe₂ hollow nanospheres. In addition, it was found that MoSe₂ has an excellent tribological property, especially coatings [34-35]. However, to the best of our knowledge, little work focused on the tribological properties of MoSe₂ as lubrication additive.

In this work, the MoSe₂ hollow nanospheres were successfully prepared by hydrothermal reaction (NH₄)₂MoO₄·2H₂O and Se powder in mixture of hydrazine hydrate and water at high temperature. The tribological properties of MoSe₂ hollow nanospheres as additives in the 100SN base oil were also investigated. This study provides a simple and novel route for the hydrothermal preparation of MoSe₂ and will be useful for MoSe₂ practical application in the future.

2. Experimental

2.1. Synthesis of MoSe₂ hollow nanospheres:

All chemical reagents were of analytic purity and applied directly without further purification. First, ammonium molybdate (NH₄)₂MoO₄·2H₂O (0.507 g), hydrazine hydrate (N₂H₄·H₂O) (10 mL), Se powder (0.316 g) were dissolved in 60 ml deionized water under constant stirring. The pH value of the mixture was adjusted to 12 by addition of NaOH solution. The final solution was transferred into a 100 ml Teflon-lined stainless steel autoclave, which was sealed and treated at 200°C for 48 h and cooled down to room temperature naturally. The resulting suspension were washed several times with distilled water and absolute ethanol, and dried in air at 70°C for 12 h.

2.2. Characterisation of MoSe₂ samples:

The phase and crystallinity of the as-synthesized products were characterized by X-ray diffraction with aD8 advance (Bruker-AXS) diffractometer employing CuK α radiation operated at 40Kv and 20mA, data analysis with Jade software. The composition was characterised by energy-dispersive spectroscopy (EDS). The morphologies and structures of the samples were investigated by scanning electron microscopy (SEM, JEOL JSM-7001F) equipped with single crystal W cathode, and transmission electron microscope (TEM, JEOL JEM-2100) equipped with LaB₆ cathode using an accelerating voltage of 200 kV. The samples for the SEM and EDS studies were prepared by placing the MoSe₂ powders on to a copper disk with conducting resin followed by metal spraying. The samples for the TEM studies were prepared by dispersing the powder in ethanol through ultra-sonication and drop casting the dispersion onto a carbon coated copper grid.

2.3. Tribological properties of MoSe₂ hollow nanospheres as a lubrication additive:

The as-prepared MoSe₂ hollow nanospheres were distributed into the 100SN base oil via 60 min ultra-sonication without any active reagent. The friction and anti wear properties of the oil with MoSe₂ samples were examined on a UMT-2 ball-on-disc friction and wear tester at ambient conditions. The testing of the friction reduction and wear resistance was conducted at a rotating speed of 300 rpm and a load of 5–30 N for 200 m. The material of the upper sample was a 440C stainless-steel ball with a diameter of 10 mm, a hardness of 62 HRC and the counterpart is a 45 steel disc of $\varnothing 30$ mm \times 5 mm in size. The friction coefficient was recorded automatically. Morphologies of wear scars were examined using a BRUKER Contour GT non-contact 3D optical profile testing instrument.

3. Results and discussions

The crystallinity, structure, and phase purity of the prepared samples were confirmed by XRD and EDS. Fig. 1a shows the XRD pattern of the MoSe₂ sample fabricated by the hydrothermal reaction (NH₄)₂MoO₄·2H₂O and Se powder in mixture of hydrazine hydrate and distilled water at 200 °C for 48 h. All peaks in the XRD pattern can be readily indexed to the hexagonal (P6₃/mmc space group) MoSe₂ phase, with lattice constants $a=3.288\text{\AA}$, $c=12.930\text{\AA}$, which are in good agreement with the reported values (JCPDS card no. 87-2419). No characteristic

peaks from other impurities are observed in the XRD pattern, indicating the high purity of the MoSe_2 samples. The XRD peaks are a little broad and this suggests that the as-prepared products are somewhat amorphous during the hydrothermal process. Energy-dispersive X-ray spectrometer (EDS) result as shown in Fig. 1b reveals that the sample consisted of element Mo and Se, no another element is observed. Furthermore, the quantification of the peaks shows that the atom ratio between Mo and Se is about 1:2.22, which is close to the stoichiometric MoSe_2 .

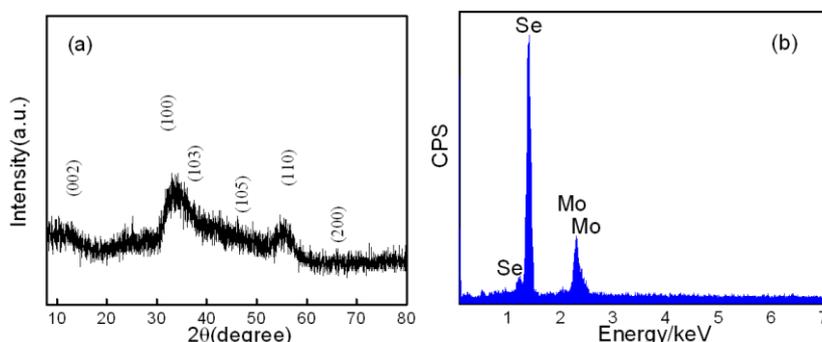


Fig. 1.(a) XRD pattern and (b)EDS of the as-prepared MoSe_2 .

The morphology of the obtained MoSe_2 products was investigated by SEM (Fig. 2a and 2b) TEM (Fig. 2c) and HRTEM (Fig. 2d). Fig. 2a displayed the low magnification SEM images of MoSe_2 nanomaterials architecture, and these nanomaterials composed of many aggregated particles. An enlarged SEM image apparently showed that the obtained samples mainly comprised of nanospheres, as shown in Fig. 2b. Further insight into the morphology and microstructure of MoSe_2 nanomaterials was gained by TEM. As shown in Fig. 2c, the low-magnification TEM imaged shows that the as-synthesized MoSe_2 nanomaterials assembled of many uniform hollow nanospheres with an average size of $\sim 50\text{nm}$, the results are consistent with the above SEM observation. The high-magnification TEM image, as shown in Fig. 2d, indicates that MoSe_2 hollow nanospheres mainly consisted of 3~ layers stacking of the monatomic sheets, the interlayer separation between the MoSe_2 layers is about 0.67nm .

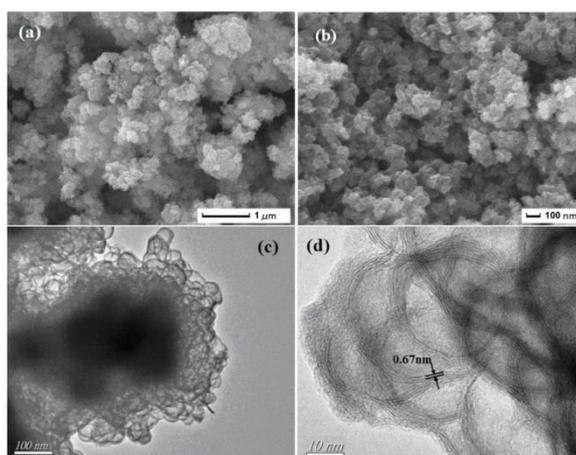


Fig. 2. FESEM (a,b), TEM(c) and HRTEM (d) image of the MoSe_2 nanospheres.

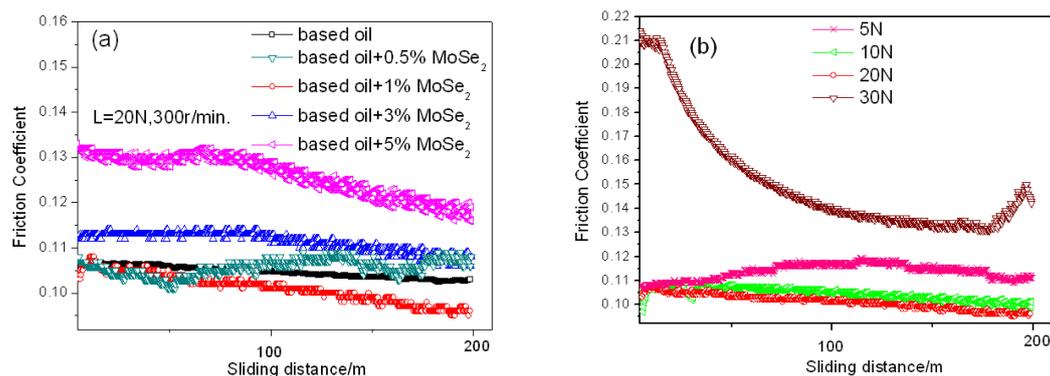


Fig. 3. Friction coefficients vs. sliding distance curves for base oil containing different concentrations MoSe₂ (a) and base oil mixed with 1.0 wt% MoSe₂ additive under different loads (b).

In order to investigate the evolutions of the friction coefficients with the different concentrations of MoSe₂, the tribological performance of the oil with different contents of MoSe₂ were tested under a condition of 20 N, 300r/min and 200m, respectively, as provided in Fig. 3a. It can be found that the base oil with 1wt% MoSe₂ gives a lower and more stable friction coefficient than others. This is not the case with the 0.5 %, 3% and 5% content of MoSe₂ additive. With regards to the 0.5wt % content of MoSe₂, the friction coefficient is decreases firstly, and then it progressively increases. With regards to the 3wt % content of MoSe₂, at the beginning of the test and up to 100m, the friction coefficient is fairly stable at ~0.113, then it progressively decreases along the sliding distance. The oil formulated with 5wt % MoSe₂ shows different distance dependence to that of with other content of MoSe₂, and with somewhat higher and unstable friction coefficient. When the concentration of synthesized MoSe₂ is 1wt%, the best friction coefficient-reducing property is obtained. The excellent tribological performance can be attributed to their hollow spheroidal and extremely thin laminated structure. Moreover, the nanoparticles can easily enter into the interface with base oil and deposit a continuous protective film, so the performance of friction coefficient is more stable than others, however, too much higher concentration of the nanoparticles exhibited extensive agglomeration, could not enter the contact area easily [36–37], also could destroy the stability of the colloid system of the base oil [38]. Therefore, the optimum concentration of the synthesized MoSe₂ as an additive in base oil is suggested to be 1wt%.

Fig. 3b show the variations of friction coefficient of 1 wt% MoSe₂hollow nanospheres as an additive of lubricant under different loads, respectively. It can be seen that there is a tendency that the friction coefficient of MoSe₂decreases firstly and then increases as the load increases, also, the curves of 5N and 30N have higher noise than the curves of 10N and 20N. We can get the optimal load for friction performance at 10N and 20 N. In a certain pressure range, MoSe₂hollow nanospheres is crushed and delaminated easily at the contact zone and form a tribofilm, which can decrease shearing stress and give a low friction coefficient. When the applied load is 5 N, MoSe₂ cannot be compacted on the worn surface to mend wear scar, so the protective film was not formed. On the contrary, if the applied load is too high, the protective film at the contact zonewill be destructed, finally leading to high noise and friction coefficient.

In order to investigate the wear resistance property of MoSe₂ hollow nanospheres as lubricant oil additives, the wear scars of plate after rubbing were tested by a BRUKER Contour GT non-contact 3D optical profile testing instrument. Fig. 4 illustrates wear scars of the base oil and oil with 1wt % MoSe₂at 300 r/min under 20 N loads. It can be clear seen that the wear scar for base oil is composed of wide grooves and irregular pits along the sliding direction (Fig 4a), and the grinding crack caused by the base oil with 1wt% MoSe₂ hollow nanospheres (Fig 4b) is shallower and smoother than that caused by the base oil (Fig 4a). From the images we can see that the depth and width of the wear scar for base oil with 1wt% MoSe₂ hollow nanospheres are about 30μm and 110μmrespectively, while those for pure base oil are about 60μm and 180 μm. This proves that the

base oil with 1wt% MoSe₂ hollow nanospheres represented better anti-wear capability than base oil.

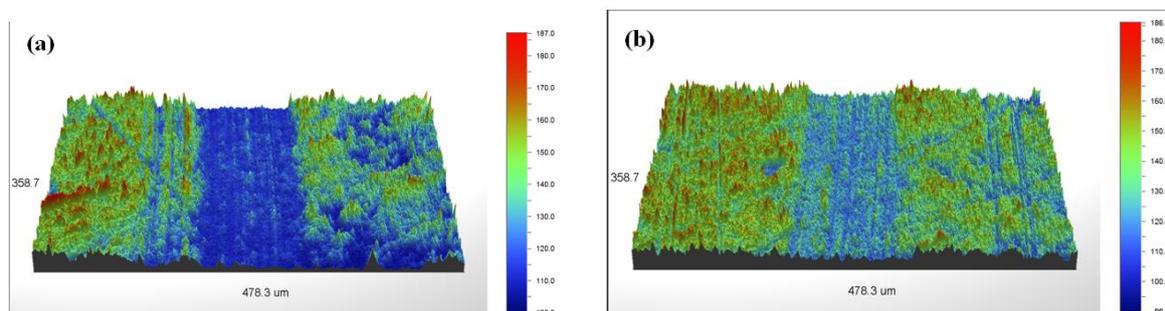


Fig.4. Non-contact 3D optical profile testing instrument images of the wear scars of (a) the base oil, (b) the base oil with 1w.t.%MoSe₂ at 300 r/min under 20 N loads for 200m

From the above results, MoSe₂ hollow nanospheres as lubrication additive could improve tribological properties of the base oil. The tribological mechanism between the rubbing surfaces in base oil with MoSe₂ hollow nanospheres is vividly proposed in Fig. 5. It is believed that the suitable content of MoSe₂ hollow nanospheres could enter in the gap between the friction surfaces effectively, which is responsible for the friction coefficient improvement. Studies have shown that the exfoliated molecular sheets from hollow nanospheres transfer onto the rough surface and form a tribofilm, which is the main mechanism for improving the tribological behavior of hollow nanospheres, while the rolling friction here also plays some role. In this experiment, MoSe₂ hollow nanospheres having a rolling antifriction temporarily at the initial stage, with the prolong of sliding time and the change of normal load, hollow nanospheres gradually deformed and began exfoliated, then the exfoliated MoSe₂nanosheetis adsorbed on the surface of the friction pair to form a layer of tribofilm, which can decrease shearing stress, and result in a decrease of the friction coefficient.

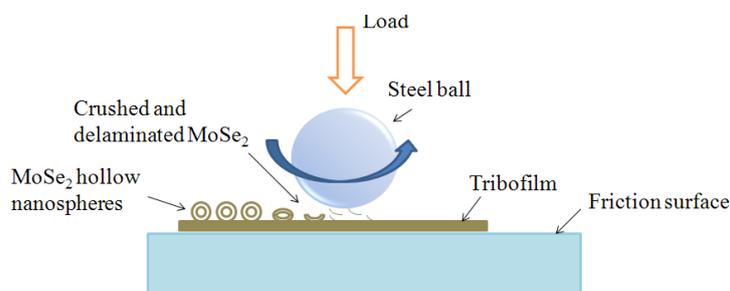


Fig. 5.Mechanism model of friction reduction and wear resistance between friction surfaces in the base oil with MoSe₂ hollow nanospheres

4. Conclusions

The novel MoSe₂ hollow nanospheres have successfully synthesized by a facile and effective hydrothermal method. The results indicated that the MoSe₂hollow nanospheres with diameter~50 nm, and consisted of 3~ layered structure MoSe₂nanosheets. The tribological measurements showed that the optimum concentration (1.0 wt%)MoSe₂hollow nanospheres by mixing with base oil has excellent friction reduction and wear resistance properties. Furthermore, the experiment results suggested that MoSe₂hollow nanospheres crushed and delaminated and formed a tribofilm on the worn surface, which could improve the tribological properties. So, MoSe₂ hollow nanospheres reveal a potential application in lubrication.

Acknowledgments

This work was supported by the Natural Science Foundation of the Jiangsu Higher Education Institutions of China (16KJB430031, 14KJB460012), 333 Project of Jiangsu Province, higher vocational college teacher training program in Jiangsu Province (2016GRFX005), Natural Science Foundation of the Changzhou Vocational Institute of Light Industry (QNJJ1403), and Innovation Training Programs for Undergraduates of Jiangsu Province (201613101006Y)

References

- [1] X Zhang, H Tang, C Lia, et al. Chalcogenide Letters **10**(10), 403 (2013).
- [2] D Kong, H Wang, J J Cha, et al. Nano letters **13**(3), 1341 (2013).
- [3] S Tongay, J Zhou, C Ataca, et al. Nano letters **12**(11), 5576 (2012).
- [4] Y Shi, C Hua, B Li, et al. Advanced Functional Materials **23**(14), 1832 (2013).
- [5] L T L Lee, J He, B Wang, et al. Scientific reports, 2014, p. 4.
- [6] C Tsai, K Chan, F Abild-Pedersen, et al. Physical Chemistry Chemical Physics **16**(26), 13156 (2014).
- [7] P Tonndorf, R Schmidt, P Böttger, et al. Optics express, **21**(4), 4908 (2013).
- [8] A Splendiani, L Sun, Y Zhang, et al. Nano letters **10**(4), 1271 (2010).
- [9] Y L Song, H H Peng, C S Li, et al. Materials for Mechanical Engineering **1**, 026 (2011).
- [10] H Li, L Chen, Y Zhang, et al. Crystal Research and Technology **49**(4), 204 (2014).
- [11] T Kubart, T Polcar, L Kopecký, et al. Surface and Coatings Technology **193**(1), 230 (2005).
- [12] S Larentis, B Fallahazad, E. Tutuc, Applied Physics Letters **101**(22), 223104 (2012),
- [13] T J S Anand, C Sanjeeviraja, M. Jayachandran, Vacuum **60**(4), 431 (2001).
- [14] J Etzkorn, H A Therese, F Rocker, et al. Advanced Materials **17**(19), 2372 (2005).
- [15] Y Zhang, Q Gong, L Li, et al. Nano Research, 2014, p.1-8.
- [16] C Fan, Z Wei, S Yang, et al. RSC Advances **4**(2), 775 (2014).
- [17] H Tang, K Dou, C C Kaun, et al. Journal of Materials Chemistry A **2**(2), 360 (2014).
- [18] J H Zhan, Z D Zhang, X F Qian, et al. Materials research bulletin **34**(4), 497 (1999).
- [19] X Wang, Y Gong, G Shi, et al. ACS nano **8**(5), 5125 (2014).
- [20] J Etzkorn, H A Therese, F Rocker, et al. Advanced Materials **17**(19), 2372 (2005).
- [21] J C Shaw, H Zhou, Y Chen, et al. Nano Research **7**(4), 511 (2014).
- [22] V G Pol, S V Pol, P George, A. PandGedanken, Journal of Materials Science **43**(6), 1966 (2008).
- [23] X Zhang, D Zhang, H Zhang, H Tang, C Li, Y. Jin, Chalcogenide Letters **11**(1), 1 (2014).
- [24] S Tongay, J Zhou, C Ataca, et al. Nano letters **12**(11), 5576 (2012).
- [25] H Li, J Wu, Z Yin, et al. Accounts of chemical research **47**(4), 1067 (2014).
- [26] J N Coleman, M Lotya, A O'Neill, et al. Science **331**(6017), 568 (2011).
- [27] Y Yao, M T McDowell, I Ryu, et al. Nano letters **11**(7), 2949 (2011).
- [28] K Tang, L Fu, R J White, et al. Advanced Energy Materials **2**(7), 873 (2012).
- [29] G Lin, J Zheng, R. Xu The Journal of Physical Chemistry C **112**(19), 7363 (2008).
- [30] C Zhu, B Lu, Q Su, et al. Nanoscale **4**(10), 3060 (2012).
- [31] C Y Cao, W Guo, Z M Cui, et al. Journal of Materials Chemistry **21**(9), 3204 (2011).
- [32] C Li, Y Liu, L Li, et al. Talanta **77**(1), 455 (2008).
- [33] J Park, X Shen, G. Wang, Sensors and Actuators B: Chemical **136**(2), 494 (2009).
- [34] T Kubart, T Polcar, L Kopecký, R Novak, D. Novakova Surface and Coatings Technology, **193**(1), 230 (2005).
- [35] T Polcar, M Evaristo, M Stueber, A. Cavaleiro, Wear **266**(3), 393 (2009).
- [36] G Tang, J Zhang, C Liu, et al. Ceramics International **40**(8), 11575 (2014).
- [37] M Zhang, B Chen, H Tang, et al. Rsc Advances **5**(2), 1417 (2014).
- [38] G Zhao, Q Zhao, W Li, X Wang, W. Liu, Lubrication Science **26**(1), 43 (2014).