

STRUCTURE AND VIBRATIONAL MODES OF Ge-S-Se GLASSES: RAMAN SCATTERING AND AB INITIO CALCULATIONS

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Amorphous Ge-S-Se glasses samples were prepared by conventional melt-quenching techniques with Ge, S and Se of 99.99% purity. Raman spectra of samples were obtained by a Fourier transform Raman spectrometer (model RM-1000, Renishaw, UK). In addition, we have used a developed computational technique based on Hartree-Fork theory to study the Raman-active modes of five tetrahedral molecular units $\text{GeS}_x\text{Se}_{4-x}$ ($x=0, 1, 2, 3, 4$), edge-sharing(ES)- Ge_2S_6 , edge-sharing(ES)- Ge_2Se_6 and ethane-like(ETH)- Ge_2S_6 , ethane-like(ETH)- Ge_2Se_6 . The positions of the calculated stretching modes in the basic cluster models are in excellent agreement with observed Raman spectra. Moreover, the basic cluster results suggest that the changes of calculated stretching modes in $\text{GeS}_x\text{Se}_{4-x}$ ($x=0, 1, 2, 3, 4$) are corresponding to vibrational coupling theory.

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1. Introduction

Chalcogenide glasses have turned out to be a very bright field of experimental and theoretical research over the last 30 years, which is the outcome of an impressive variety of athermal photoinduced phenomena [1]. Among the most promising chalcogenide glasses, great attention has focused on the Ge-S and Ge-Se glasses due to their interesting physical properties. So, it is essential for us to understand details of short-range structural order of chalcogenide glasses in order to advance structure-properties correlations [2]. Raman spectroscopy has been considered to be a significant and powerful tool for investigating the structure of short-range order and network rigidity in chalcogenide glasses [3, 4, 5]. Meanwhile, the Raman spectra is a simple and direct probe of the molecular structure of chalcogenide glass alloys, which are particularly easy to interpret [6, 7]. At present, there are several methods of investigation, which have been used to acquire insight into the structural properties of chalcogenide glassy systems such as ab initio method of Hartree-Fork theory [2], density functional theory (DFT) [8] and method of analogy between clusters, which have similar microstructure and reduced mass [9].

In this paper, Amorphous $\text{Ge}(\text{S}_x\text{Se}_{1-x})_2$, $x=0, 0.2, 0.4, 0.5, 0.6, 0.8, 1.0$ were prepared by conventional melt-quenching techniques. Then, the Hartree-Fork (HF) molecular orbital theory in Gaussian03 program package was used to calculate the basic structural units and the vibrational frequencies of four molecular units $\text{GeS}_x\text{Se}_{4-x}$ ($x=0, 1, 2, 3, 4$), ES- Ge_2S_6 , ES- Ge_2Se_6 and ETH- Ge_2S_6 , ETH- Ge_2Se_6 .

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2. Experimental

Samples of amorphous $\text{Ge}(\text{S}_x\text{Se}_{1-x})_2$ ($x=0,0.2,0.4,0.5,0.6,0.8,1.0$) were prepared by conventional melt-quenching techniques with Ge, S and Se of 99.99% purity. Afterwards, those powders with composition above were grounded and weighed exactly in appropriate quantity and fused quartz ampoules containing the raw powders were put in evacuated (Vacuum: 10^{-1} Pa) and then inserted into a rocking furnace. The batches were held at 450°C for about 1 hour in order to promote the reaction completely among Ge, S and Se and subsequently melted in a rocking furnace at 950°C for 12-18 hours. With a view to stabilizing the melts for homogeneity without rotation, we keep placing for 1 hour with close of the rock and these samples were quenched in ice water for a few seconds.

To study the Raman-active modes of $\text{GeS}_x\text{Se}_{4-x}$, we use finite clusters of atoms containing structural units, which are on the basis of central-force network model reported by Frank L. Galeener [10]. The research has demonstrated that effective coupling of neighboring tetrahedral is zero when the angle of A-X-A for AX_2 glasses and vibrational modes are determined more by the isolated AX_4 tetrahedra rather than by the A-X-A bonds when the average A-X-A bond angle is smaller than the critical angle $\cos^{-1}(-2m/3M)$, where M is the mass of the A atom and m is that of the X atom [11]. We can obtain the critical angle values of Ge-S-Ge and Ge-Se-Ge are 107.11° and 136.44° , respectively. In addition, with the calculation of molecular dynamics simulations in glassy GeS_2 [12] and the analysis of neutron and high-energy X-ray diffraction measurement in glassy GeSe_2 [13], it can be easily found that the average angle values of Ge-S-Ge and Ge-Se-Ge are smaller than the critical angle value, so the glassy system can be considered to be isolated tetrahedral for the sake of calculating the vibrational property of short-range structure.

According to the research reported in 1983, J.E.Griffiths[9] have measured the Raman spectra of bulk $\text{Ge}(\text{S}_x\text{Se}_{1-x})_2$ glasses and molecular of $\text{GeCl}_x\text{Br}_{4-x}$, which predicted the glass of $\text{Ge}(\text{S}_x\text{Se}_{1-x})_2$ may be expected consist of linked tetrahedral of $[\text{GeSe}_4]$, $[\text{GeSe}_3\text{S}]$, $[\text{GeSe}_2\text{S}_2]$, $[\text{GeSeS}_3]$ and $[\text{GeS}_4]$ in diminishing proportions analogous to molecular $[\text{GeBr}_4]$, $[\text{GeBr}_3\text{Cl}]$, $[\text{GeBr}_2\text{Cl}_2]$, $[\text{GeBrCl}_3]$ and $[\text{GeCl}_4]$. So it is reasonable to calculate the vibrational modes of $\text{GeCl}_x\text{Br}_{4-x}$ in order to approach the vibrational modes of $\text{GeS}_x\text{Se}_{4-x}$. On the other hand, some researches [10] show the hydrogen atoms in the cluster are used as pseudoatoms to replace the solid glass by finite molecular model, because the difference of large mass between H and the heavy atoms can prevent H-atom motion from mixing strongly in the Ge-X ($x=\text{S,Se}$) vibrational modes and any H-atom modes that do fall in this region are removed from the analysis of region for GeS_2 or GeSe_2 [3].

Calculations reported in this work were performed with the Gaussian 03 and Gaussview 5.0 program package. And the Hartree-Fork molecular orbital theory was utilized in the present study in order to calculate the basic structural details and vibrational frequencies of five molecular units $\text{GeS}_x\text{Se}_{4-x}$ ($x=0, 1, 2, 3, 4$) ES- Ge_2S_6 、ES- Ge_2Se_6 and ETH- Ge_2S_6 、ETH- Ge_2Se_6 with the finite molecular model method. The geometrical optimizations of the basic structures above under the ground state were at the UHF/LanL2DZ level, which was widely used to calculate the equilibrium geometries and spectroscopic properties of clusters, particularly in the investigation on clusters with heavy atoms.

3. Results

3.1 Analogy method

Table 1, Table 2, Table 3 show the Raman vibrational frequencies and Raman activity of $[\text{GeS}_x\text{Se}_{4-x}]$, which are calculated by the data of $[\text{GeCl}_x\text{Br}_{4-x}]$ plus the similarity factor of 0.921 ($x=0, 1, 2, 3, 4$). From this three tables, we can see that there are four basic vibrational modes in $[\text{GeS}_4]$ and $[\text{GeSe}_4]$, which have the point group of T_d . When one S atom is replaced by Se atom, the point group of molecule change from T_d to C_{3v} and the basic vibrational modes increase from four to six. Then two S atoms are replaced by Se atoms, the point group of molecule change from C_{3v} to C_{2v} , and the basic vibrational modes increase from six to nine. In addition, ν_1 -type mode in Table 1 is attributed to the symmetrical stretching vibration, however, ν_3 -type mode is attributed to antisymmetric stretching vibration. Also we can find that the frequencies of ν_3 are higher than those of ν_1 , which is accordance with the articles published past [14]. According to the Raman

activity value, we can easily find the main Raman vibrational mode in each molecular unit.

Table 1. The Raman vibrational frequencies (cm^{-1}) and Raman activity (A^4 , AMU) of molecule with T_d point group

T_d	$\nu_1(A_1)$	$\nu_2(E)$	$\nu_3(F_2)$	$\nu_4(F_2)$
XY_4	$\nu_s(XY)$	$\delta_d(YXY)$	$\nu_8(XY)$	$\delta_d(YXY)$
GeS ₄	340	116	417	153
Scattering Activity	41.45	6.47	9.50	6.53
GeSe ₄	198	71	295	101
Scattering Activity	39.41	6.08	11.20	7.30

Table 2. The Raman vibrational frequencies (cm^{-1}) and Raman activity (A^4 , AMU) of molecule with C_{3v} point group

C_{3v}	$\nu_1(A_1)$	$\nu_2(A_1)$	$\nu_3(A_1)$	$\nu_4(E)$	$\nu_5(E)$	$\nu_6(E)$
ZXY_3	$\nu(XY_3)$	$\nu(XZ)$	$\delta(XY_3)$	$\nu(XY_3)$	$\delta_s(XY_3)$	$\rho_r(XY_3)$
GeS ₃ Se	374	270	107	410	143	102
Scattering Activity	21.85	26.24	8.62	11.06	6.37	6.29
GeSSe ₃	218	388	110	277	122	79
Scattering Activity	35.40	17.10	7.05	9.50	7.54	6.20

Table 3. The Raman vibrational frequencies (cm^{-1}) and Raman activity (A^4 , AMU) of molecule with C_{2v} point group

C_{2v}	$\nu_1(A_1)$	$\nu_2(A_1)$	$\nu_3(A_1)$	$\nu_4(A_1)$	$\nu_5(A_2)$	$\nu_6(B_1)$	$\nu_7(B_1)$	$\nu_8(B_2)$	$\nu_9(B_2)$
Z_2XY_2	$\nu(XY)$	$\nu(XZ)$	$\delta(XY_2)$	$\delta(XZ_2)$	$\rho_r(XY_2)$	$\nu(XY)$	$\rho_m(XY_2)$	$\nu(XZ)$	$\rho_r(XY_2)$
GeS ₂ Se ₂	384	241	136	84	95	403	124	308	132
Scattering Activity	19.16	31.04	7.98	6.48	5.97	12.50	6.20	7.82	8.01

3.2 Direct computing by terminate H atom

In order to study the Raman-active modes of GeS_xSe_{4-x}, the finite clusters of atoms should be obtained containing structural units expected to be significant in the glasses. Dangling bonds on the cluster surfaces are terminated by H atoms, which provide a better model the chemical environment of the glasses. Because of the large mass difference between H and the heavy atoms(Ge、S、Se),the vibrational modes related to H-atoms lie well outside the spectral region of Ge-S or Ge-Se vibrational modes. It is easy to remove any H-atom modes that do fall in this region from the analysis. All vibrational modes can be attributed as follows [8]:

Table 4. The Raman vibrational frequencies (cm⁻¹) and Raman activity (A⁴, AMU) of molecule with T_d point group

T _d	v ₁ (A ₁)	v ₂ (E)	v ₃ (F ₂)	v ₄ (F ₂)
XY ₄	v _s (XY)	δ _d (YXY)	v _δ (XY)	δ _d (YXY)
GeS ₄	332	123	416	153
Scattering Activity	77.16	2.30	20.19	11.47
GeSe ₄	194	68	295	98
Scattering Activity	48.70	9.91	13.93	8.40

Table 5. The Raman vibrational frequencies (cm⁻¹) and Raman activity (A⁴, AMU) of molecule with C_{3v} point group.

C _{3v}	v ₁ (A ₁)	v ₂ (A ₁)	v ₃ (A ₁)	v ₄ (E)	v ₅ (E)	v ₆ (E)
ZXY ₃	v(XY ₃)	v(XZ)	δ(XY ₃)	v(XY ₃)	δ _s (XY ₃)	ρ _r (XY ₃)
GeS ₃ Se	367	265	141	395	158	110
Scattering Activity	40.32	42.88	12.95	21.36	17.68	3.66
GeSSe ₃	208	392	111	303	113	90
Scattering Activity	45.21	31.42	9.04	17.03	9.04	2.51

Table 6. The Raman vibrational frequencies (cm⁻¹) and Raman activity (A⁴, AMU) of molecule with C_{2v} point group.

C _{2v}	v ₁ (A ₁)	v ₂ (A ₁)	v ₃ (A ₁)	v ₄ (A ₁)	v ₅ (A ₂)	v ₆ (B ₁)	v ₇ (B ₁)	v ₈ (B ₂)	v ₉ (B ₂)
Z ₂ XY ₂	v(XY)	v(XZ)	δ(XY ₂)	δ(XZ ₂)	ρ _r (XY ₂)	v(XY)	ρ _m (XY ₂)	v(XZ)	ρ _r (XY ₂)
GeS ₂ Se ₂	383	236	127	82	102	403	120	309	125
Scattering Activity	33.98	40.07	11.88	13.58	2.35	24.95	5.26	11.97	9.33
Intensity	2.5	3.6	1.5	1.3	0.3	1.9	0.7	1.9	1.4

Apart from the molecule of GeS_xSe_{4-x}, ES-Ge₂S₆, ES-Ge₂Se₆ and ETH-Ge₂S₆, ETH-Ge₂Se₆ are also important basic units in the Ge-S-Se glass system, So we should pay more attention about them. According to the calculation by Gaussian software, there are 12 basic vibrational modes in ETH-Ge₂S₆ and ETH-Ge₂Se₆. In addition, there are 18 basic vibrational modes in ES-Ge₂S₆ and ES-Ge₂Se₆. the data are as follows:

Table 7. The basic calculated vibrational modes of *ETH-Ge₂S₆* and *ETH-Ge₂Se₆*.

Vibrational mode	ETH-Ge ₂ S ₆	ETH-Ge ₂ Se ₆
v ₁	373	299
v ₂	257	176
v ₃	110	66
v ₄	14	8
v ₅	346	218
v ₆	149	110
v ₇	394	295
v ₈	164	129
v ₉	116	98
v ₁₀	400	291
v ₁₁	155	100
v ₁₂	65	44

Table 8. The main vibrational modes of *ETH-Ge₂S₆* and *ETH-Ge₂Se₆*

D _{3d}	v ₁ (A _{1g})	v ₂ (A _{1g})
X ₂ Y ₆	v _s (XY ₃)	v(XX)
Ge ₂ S ₆	373	257
Ge ₂ Se ₆	299	176

Table 9. The basic calculated vibrational modes of *ES-Ge₂S₆* and *ES-Ge₂Se₆*

Vibrational mode	ES-Ge ₂ S ₆	ES-Ge ₂ Se ₆
v ₁	431	315
v ₂	363	219
v ₃	171	144
v ₄	205	190
v ₅	68	44
v ₆	337	244
v ₇	136	99
v ₈	421	308
v ₉	31	15
v ₁₀	187	113
v ₁₁	416	306
v ₁₂	124	79
v ₁₃	398	286
v ₁₄	112	65
v ₁₅	166	78
v ₁₆	406	294
v ₁₇	328	190
v ₁₈	132	86

Table 10. The main vibrational modes of ES-Ge₂S₆ and ES-Ge₂Se₆

D _{2h}	v ₁ (A _{1g})	v ₂ (A _{1g})
X ₂ Y ₆	v(XY)	v(XY)
Ge ₂ S ₆	431	363
Ge ₂ Se ₆	315	219

4. Discussion

Table 11 shows the Raman main vibrational mode of GeS_xSe_{4-x} (x=0, 1, 2, 3, 4) between calculated data and experimental data. We can see that the Raman main vibrational modes calculated by Gaussian are in accordance with those of experimental data. However, from the table, not all main vibrational modes of GeS_xSe_{4-x} are v₁-type modes, which is quite different from the research of J.E.Griffith in his paper published in the year 1983, where they considered that all main vibrational modes of GeS_xSe_{4-x} are v₁-type, but we can see the Raman main vibrational modes of [GeS₃Se] and [GeS₂Se₂] are v₂-type.

Table 11 also shows the frequency variation of Raman main vibrational modes in different basic units. For example, the frequency of the Raman main vibrational mode of [GeS₄] is about 340cm⁻¹, however, the frequency decreases sharply till 270cm⁻¹, when one S atom is replaced by Se atom. Differential between [GeS₄] and [GeS₃Se] is approximately 70cm⁻¹, but the differential between [GeS₃Se] and [GeS₂Se₂] is about 29cm⁻¹, which is similar to the differential between [GeS₂Se₂] and [GeSSe₃] and the differential between [GeSSe₃] and [GeSe₄]. According to the analysis of vibrational mode and the data in table, only the main vibrational mode of [GeS₄] in this five basic units is related to Ge-S stretching vibration, and others are all related to Ge-Se stretching vibration. This is because the atomic mass of Se is heavier than that of S, and the electron cloud of Se atom change easier than that of S atom. So the Raman activity of Ge-Se bond is larger than that of Ge-S bond, and the interaction force of Ge-Se is smaller than that of Ge-S. Based on the molecular spectroscopy theory as follows:

$$v \propto \sqrt{\frac{f}{\mu}}$$

Where f is a constant related to the bond strength, and μ is the discount mass. So when S atoms are replaced by Se atoms, value of f is smaller and μ is bigger, which leads to a decrease of frequency. Fig1 demonstrates the calculated Raman spectroscopy of GeS_xSe_{4-x} by Gaussian03. We can easily find that the highest peak of each tetrahedral unit moves to lower frequency with the increase of Se atoms.

Table 12 illustrates the Ge-S symmetrical stretching vibrations of GeS_xSe_{4-x}, however, Table 13 shows the Ge-Se symmetrical stretching vibrations of GeS_xSe_{4-x}. From the data in these two tables, the law of frequency shifts related to Ge-S bond and Ge-Se bond are similar. For example, there is one Ge-S bond in the molecule of [GeSSe₃] and the frequency of Ge-S stretching vibrational mode is 392cm⁻¹. When the Se atom is replaced by S atom in [GeS₂Se₂], the stretching vibration of two Ge-S bonds in [GeS₂Se₂] produce coupling, which are divided into two vibrational modes. One is called symmetrical stretching vibration [v₁(A₁)], the other is called asymmetrical stretching vibration [v₆(B₁)]. The frequencies are 383cm⁻¹ and 403cm⁻¹ respectively. When there is one Se atom replaced by S atom in [GeS₂Se₂], the stretching vibration of three Ge-S bonds in [GeS₃Se] produce coupling, which are also divided into two vibrational modes. One is called symmetrical stretching vibration [v₁(A₁)], the other is called asymmetrical stretching vibration [v₄(E)]. The frequencies are 367cm⁻¹ and 395cm⁻¹ respectively. When the final Se atom is replaced by S atom in [GeS₃Se], the stretching vibration of four Ge-S bonds in [GeS₄] produce coupling, which are divided into two vibrational modes. One is called symmetrical stretching vibration [v₁(A₁)], the other is called asymmetrical stretching vibration [v₃(F₂)]. The frequencies

are 340cm^{-1} and 407cm^{-1} respectively. These phenomena can be explained by vibrational coupling theory. For instance, there is only one Ge-S bond in the molecule of $[\text{GeSSe}_3]$, the Ge-S bond in the $[\text{GeSSe}_3]$ is not coupled. However, the coupling of stretching vibration produces when there is more than one Ge-S bond in the molecule of $\text{GeS}_x\text{Se}_{4-x}$, which divided into symmetrical vibrational mode and asymmetrical vibrational mode. Generally speaking, the frequency of asymmetrical vibrational mode is higher than that of symmetrical vibrational mode in the same molecule. So the frequencies of symmetrical vibrational mode decrease with the increase of number of Ge-S bond. Also, the vibrational frequencies shift related to Ge-Se bond can be interpreted by vibrational coupling theory, which is the same as the explanation of Ge-S bond.

Table 11. The Raman main vibrational mode between calculated data and experimental data

Type	Main vibrational mode	Calculated (cm^{-1})	Experimental (cm^{-1})
GeS_4	$\nu_1(\text{A}_1)$ $\nu_s(\text{GeS})$	340	340
GeS_3Se	$\nu_2(\text{A}_1)$ $\nu(\text{GeSe})$	270	248
GeS_2Se_2	$\nu_2(\text{A}_1)$ $\nu(\text{GeSe})$	241	232
GeSSe_3	$\nu_1(\text{A}_1)$ $\nu(\text{GeSe}_3)$	218	218
GeSe_4	$\nu_1(\text{A}_1)$ $\nu_s(\text{GeSe})$	198	203

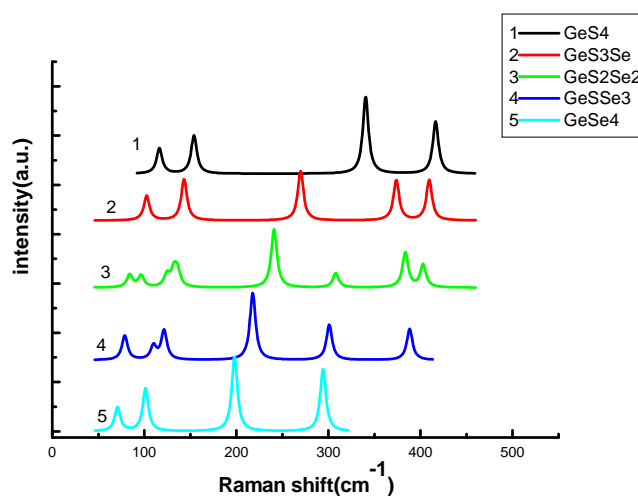


Fig 1. the calculated Raman spectroscopy of $\text{GeS}_x\text{Se}_{4-x}$ by Gaussian03

Table 12. The Ge-S symmetrical stretching vibrations of $\text{GeS}_n\text{Se}_{4-n}$ calculated by Gaussian software

Type	Ge-S symmetrical stretching vibration	Calculated (cm^{-1})
GeS_4	$\nu_1(\text{A}_1)$ $\nu_s(\text{GeS})$	340
GeS_3Se	$\nu_1(\text{A}_1)$ $\nu(\text{GeS}_3)$	374
GeS_2Se_2	$\nu_1(\text{A}_1)$ $\nu(\text{GeS})$	384
GeSSe_3	$\nu_2(\text{A}_1)$ $\nu(\text{GeS})$	388

Table 13. The Ge-Se symmetrical stretching vibrations of GeS_xSe_{4-x} calculated by Gaussian software

Type	Ge-Se Symmetric stretching vibration mode		Calculated (cm^{-1})
$GeSe_4$	$\nu_1(A_1)$	ν_s (GeSe)	198
$GeSSe_3$	$\nu_1(A_1)$	ν (GeSe ₃)	218
GeS_2Se_2	$\nu_2(A_1)$	ν (GeSe)	241
GeS_3Se	$\nu_2(A_1)$	ν (GeSe)	270

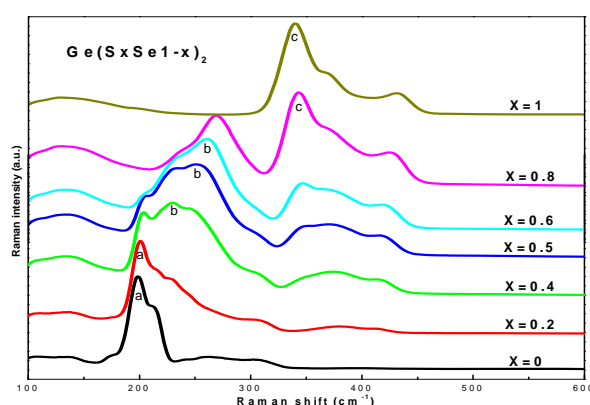
Fig 2 Raman spectroscopy of $Ge(S_xSe_{1-x})_2$ glass system.

Fig 2 shows the Raman spectroscopy of $Ge(S_xSe_{1-x})_2$ glass system, and $x=0, 0.2, 0.4, 0.5, 0.6, 0.8, 1.0$. It can be seen that there are two obviously range of peak in the whole area of Raman spectroscopy, which are located in $180-300cm^{-1}$ and $300-450cm^{-1}$ respectively. According to a paper published in the past [15] and data calculated by Gaussian, the two main peak areas can be attributed to the stretching vibrational mode related to Ge-Se bond and related to Ge-S bond respectively. With the decrease of percentage of S in $Ge(S_xSe_{1-x})_2$ system, the intensity of main peak in the Raman spectroscopy reduces and the peak in stretching vibrational area disappear when the percentage of S is close to 0. At the same time, the peaks in the stretching vibrational area related to Ge-Se move to lower frequency with the higher intensity.

The Raman spectroscopy of GeS_2 and $GeSe_2$ system are very simple compared with that of $Ge(S_xSe_{1-x})_2$ system. So we can analyze these two systems first. According to T.G.Edwards [16], the shoulder around $180cm^{-1}$ in $GeSe_2$ Raman spectroscopy is due to the stretching vibration related to Ge-Ge bond, which is accordance with the calculated data that attributed to the $\nu_3(A_{1g})$ mode in ES- Ge_2Se_6 . And the main Raman peak of $GeSe_2$ in $197cm^{-1}$ are assigned to the symmetrical stretching vibration of $[GeSe_4]$ unit, which is attributed to the $\nu_1(A_1)$ mode. In addition, the shoulder in $214cm^{-1}$ is due to the vibrational mode of the two inner Se in the dimer in ES- Ge_2Se_6 , which can be attributed to the $\nu_2(A_{1g})$ mode. The peak around $310cm^{-1}$ is mainly attributed to the $\nu_1(A_{1g})$ mode of ES- Ge_2Se_6 and it is also partially assigned to the $\nu_3(F_2)$ mode of $[GeSe_4]$ unit. On the other hand, the main Raman peak of GeS_2 system in $340cm^{-1}$ is related to the symmetrical stretching vibration of $[GeS_4]$ unit with $\nu_1(A_{1g})$ mode [17]. The shoulder of the main Raman peak in $370cm^{-1}$ is assigned to the $\nu_2(A_{1g})$ mode in ES- Ge_2S_6 [18], moreover, the peak around $430cm^{-1}$ is mainly attributed to the $\nu_1(A_{1g})$ mode of ES- Ge_2S_6 and partially assigned to the $\nu_3(F_2)$ mode of $[GeS_4]$ unit.

Apart from GeS_2 and $GeSe_2$ system, the Raman spectroscopies of $Ge(S_xSe_{1-x})_2$ glass system are a bit complicated. It can be seen from Fig 2 that the relative intensity change of the

main peak is consistent with the change of the relative ratio of S/Se in these glasses. According to the calculated data, there are five possible $\text{GeS}_x\text{Se}_{4-x}$ units with $n=0, 1, 2, 3, 4$. Among them, four units can contribute to stretching vibrational mode related to Ge-Se bond, namely $[\text{GeS}_3\text{Se}]$, $[\text{GeS}_2\text{Se}_2]$, $[\text{GeSSe}_3]$ and $[\text{GeSe}_4]$. With increasing Se in the units, the frequency would be expected to be lower. At the same time, the relative intensity in Ge-Se vibration area are higher. However, the relative intensity in Ge-S vibration area is lower. The peak around 265cm^{-1} is due to $\nu_2(A_1)$ mode in $[\text{GeS}_3\text{Se}]$ unit. In addition, the shoulder about 230cm^{-1} is attributed to $\nu_2(A_1)$ mode in $[\text{GeS}_2\text{Se}_2]$. Finally, the peak around 218cm^{-1} is assigned to $\nu_1(A_1)$ mode in $[\text{GeSSe}_3]$.

5. Conclusion

In conclusion, ab initio Raman vibrational modes calculations were carried out on basic units of Ge-S-Se glasses. It can be shown that change of point group and symmetry corresponds to the ratio of S/Se in $\text{GeS}_x\text{Se}_{4-x}$ and not all main Raman vibrational modes are attributed to ν_1 , when compared the results of vibrational frequencies and modes in $\text{GeS}_x\text{Se}_{4-x}$. Furthermore, Raman vibrational modes of $\text{GeS}_x\text{Se}_{4-x}$ are 340cm^{-1} , 248cm^{-1} , 232cm^{-1} , 218cm^{-1} , 203cm^{-1} and the shifts of vibrational frequencies are consistent with the change from stretching vibration of Ge-S to stretching vibration of Ge-Se. Finally, the stretching vibrational modes of $\text{GeS}_x\text{Se}_{4-x}$ can be divided in Ge-S relation vibration and Ge-Se relation vibration, which are consistent with vibrational coupling theory.

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